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Paper 02 - Structured Essays

Question 1 Syllabus

References: A: 7.1, 7.7,
7.3, 7.4; B2 7.1, 7.2, 7.3

Part (a) (i) tested candidates' knowledge of experimental procedure for collecting and measuring the volume of gas generated during a chemical reaction. Candidates were required to draw a suitable

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This question was attempted by majority of the candidates and the performance was fair. In (a) (i) (ii) and (iii) most candidates correctly define standard electrode potential, state factors that affect standard electrode potential and gave uses of the values of standard potential.

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Notes A is solution
containing 6.22 g of an acid

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H₂Y per dm³. B contains 3.90 g of NaOH per dm³ of solution. (a) Put A into the burette and titrate it against 20.0 cm³ or 25.0 cm³ portions of B using methyl orange as indicator. Repeat the titration to obtain consistent titres. Tabulate your

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