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Atomic Structure: Atoms consist of protons and neutrons in the nucleus, surrounded by electrons that

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reside in orbitals. Orbitals are classified according to the four quantum numbers that represent any one particular orbital energy, shape, orientation, and the spin of the occupying electron. Periodic Table: This is a tabular arrangement of the chemical elements, ordered...

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Elements are arranged on the Periodic table in order of increasing atomic number where each element has one proton more than the element preceding it. Hydrogen has 1 proton, helium has 2 protons, lithium

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has 3 etc. The table is arranged in vertical columns called Groups numbered I - VIII and in rows called Periods.

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on the periodic table is a function of their atomic structure. The periodic table is a tool used for the investigations of d) families or groups; e) periods; f) trends including atomic radii, electronegativity, shielding effect, and ionization energy; g) electron configurations, valence electrons, and oxidation

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numbers.

~~Atomic Structure: Periodic Table~~

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In the periodic table, chemical elements are arranged

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by order of atomic number in such a way that the periodic properties (chemical periodicity) of the elements are made clear. The standard form of the table includes periods (usually horizontal in the periodic table) and groups (usually vertical). Elements in groups have some similar properties.

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columns and rows they are aligned in. Each element on the periodic table has a unique atomic number which represents the number of protons the element has.

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